

CERTIFICATE OF ANALYSIS

PRODUCT : DILTIAZEM HCl (MICRONISED)		COMPENDIA : Ph.Eur.-8.0
BATCH NO. : DIL/M-01515		QUANTITY : 500 Kg
MFG. DATE : JANUARY.2015		EXP. DATE : DECEMBER.2019
QC A.R. No.: FP/022/2015		
TESTS	RESULTS	SPECIFICATIONS
01. CHARACTERS		
i. Appearance	White, crystalline powder.	White or almost white, crystalline powder.
ii. Solubility	Freely soluble in water, in methanol and in methylene chloride, slightly soluble in anhydrous ethanol.	Freely soluble in water, in methanol and in methylene chloride, slightly soluble in anhydrous ethanol.
iii. Melting point	Melted at 212°C with decomposition.	Melts at about 213° C with decomposition.
02. IDENTIFICATION		
IR Absorption	Concordant with working standard	IR Absorption spectrum of the sample is concordant with that of the Diltiazem hydrochloride CRS / working standard. It gives reaction to chlorides. Clear and colourless.
CHLORIDES	Positive	
03. APPEARANCE OF SOLUTION	Clear and colourless	
04. pH	4.9	4.3 to 5.3
05. SPECIFIC OPTICAL ROTATION	+119°	+ 115° to + 120°
06. RELATED SUBSTANCES		
i. Impurity F	Not Quantified	Not more than 0.3% w/w
ii. Unspecified Impurities	Not Quantified	Not more than 0.10% w/w each
iii. Total Impurities	Not Quantified	Not more than 0.30% w/w
07. HEAVY METALS	Less than 10 ppm	Not more than 10 ppm
08. LOSS ON DRYING	0.25%	Not more than 0.50% w/w
09. SULPHATED ASH	0.05%	Not more than 0.10% w/w
10. ASSAY (By Titration) (On dried basis)	100.01%	Not less than 98.50% w/w and Not more than 101.00% w/w
ADDITIONAL TEST:		
PARTICLE SIZE		
100%	35 Microns	100% Less than 100 Microns
Surface Area	1.73 m ² /gm	For Information
RESIDUAL SOLVENTS (BY GC-HS)		
Toluene	182 ppm	Not more than 750 ppm
Methanol	101 ppm	Not more than 500 ppm
Isopropyl Alcohol	10 ppm	Not more than 1000 ppm
Ethyl Acetate	0 ppm	Not more than 500 ppm
Bulk Density	0.28 g/ml	For Information
Tapped Density	0.48 g/ml	For Information

REMARKS: The product conforms to **Ph.Eur.-8.0 & Customer's Specifications**

Deew
Prepared by
Executive-QC

Day
Checked by
Manager-QC

Day
Approved by
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